

Formulas of Compounds

Give the correct molecular formula:

Mercury (I) cyanide _____

Radon tetroxide _____

Carbon tetrachloride _____

Lithium nitride _____

Sodium hypochlorite _____

Iron (II) chloride _____

Tetraphosphorus hexoxide _____

Nitrogen dioxide _____

Platinum (IV) phosphide _____

Aluminum bromide _____

Dinitrogen monoxide _____

Iodine pentabromide _____

Gold (III) sulfide _____

Sulfur trioxide _____

Magnesium oxide _____

Zinc acetate _____

Beryllium bicarbonate _____

Diarsenic pentasulfide _____

Potassium permanganate _____

Bismuth (III) perchlorate _____

Ammonium dichromate _____

Phosphorus trifluoride _____

Strontium fluoride _____

Barium nitrate _____

Sodium sulfite _____

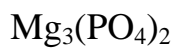
Naming Compounds

Name the following compounds:



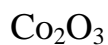












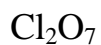


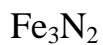












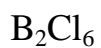
























Names & Formulas of Acids

	Anion Name	Acid Name
H_2SO_3	_____	_____
HNO_2	_____	_____
HClO_3	_____	_____
H_2S	_____	_____
HBr	_____	_____
HBrO_2	_____	_____
H_2SO_4	_____	_____
HNO_3	_____	_____
HF	_____	_____

	Anion Formula	Acid Formula
Hypobromous Acid	_____	_____
Acetic Acid	_____	_____
Phosphoric Acid	_____	_____
Hydrocyanic Acid	_____	_____
Nitric Acid	_____	_____
Hydrochloric Acid	_____	_____
Phosphorous Acid	_____	_____
Perchloric Acid	_____	_____
Carbonic Acid	_____	_____
Sulfuric Acid	_____	_____